

Preparative Chromatography with Supercritical Fluids - Operation and Optimisation



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2 Equipment and methods

Preparative Batch-SFC

Preparative SMB-SFC

Modeling

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Determination of adsorption equilibrium

Simulation / Optimization

Analysis of profitability

4 Summary

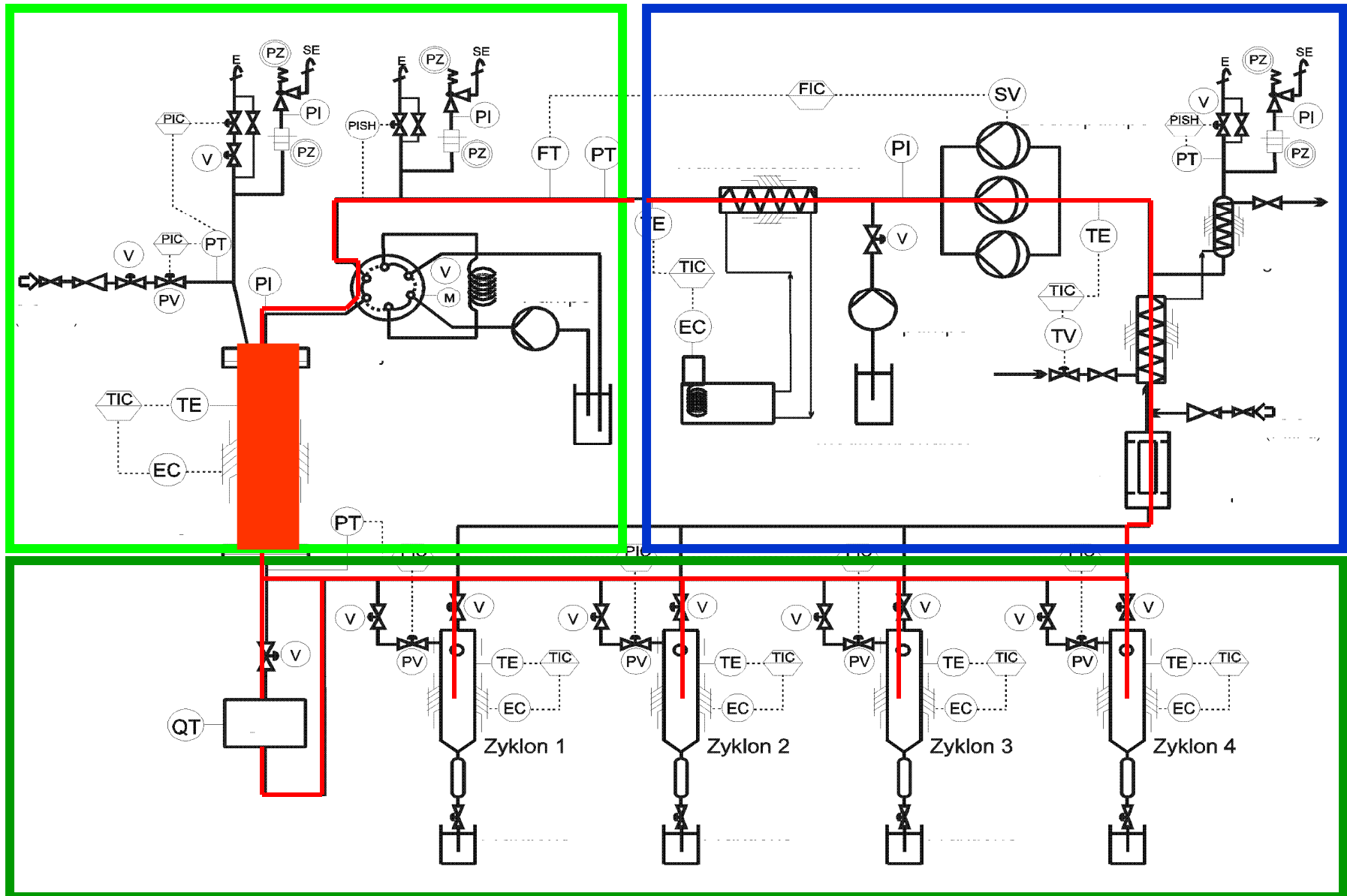
Preparative chromatography

- ➔ **Process with high selectivities** due to large number of stationary and mobile phase combinations
- ➔ For the separation of **high value products** for highest purity

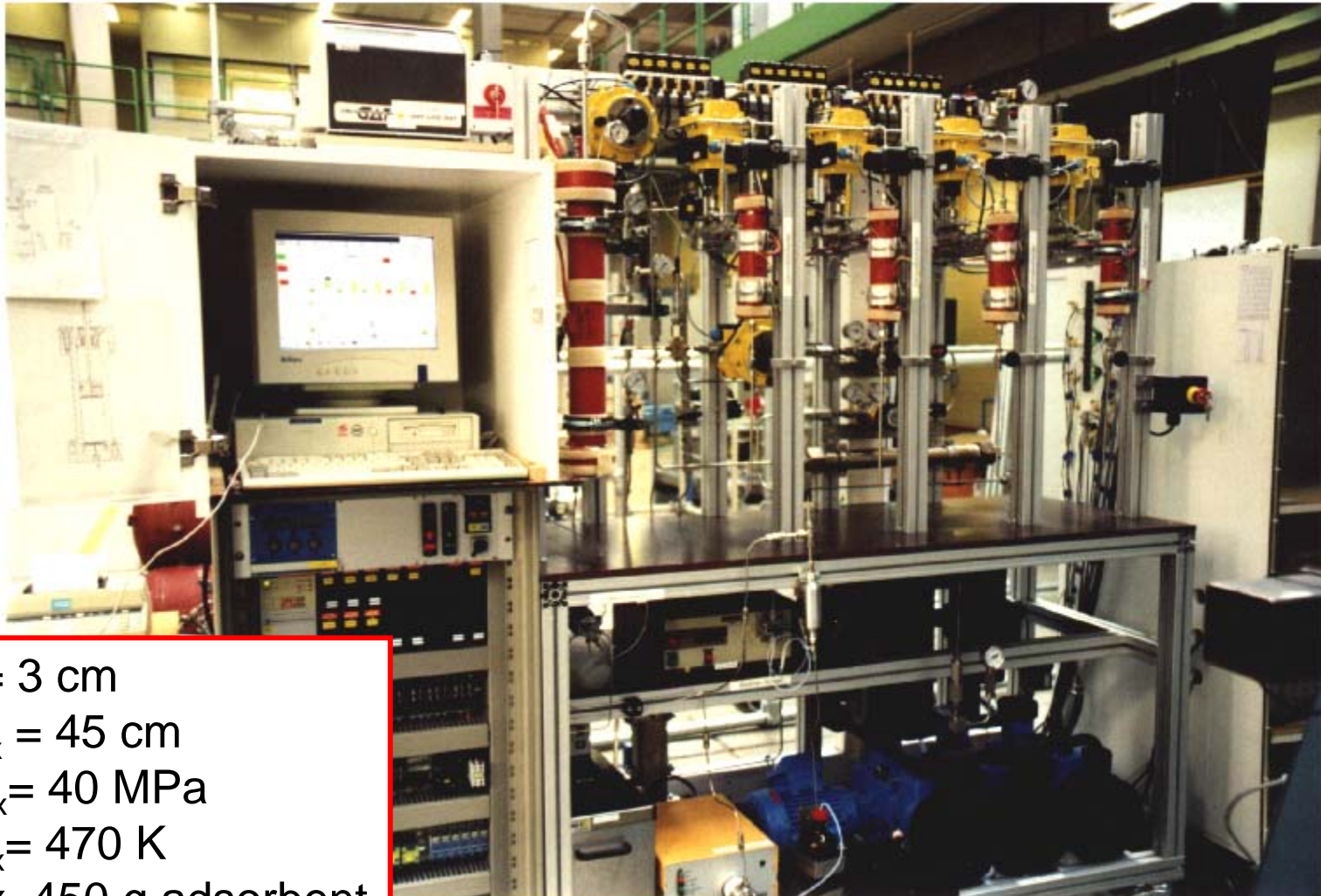
- Techniques:**
- Liquid chromatography (LC/HPLC)
 - Supercritical fluid chromatography (SFC)
 - Gas chromatography (GC)

- Modes:**
- Batch-chromatography
 - Simulated Moving Bed (SMB)

Preparative Batch-SFC

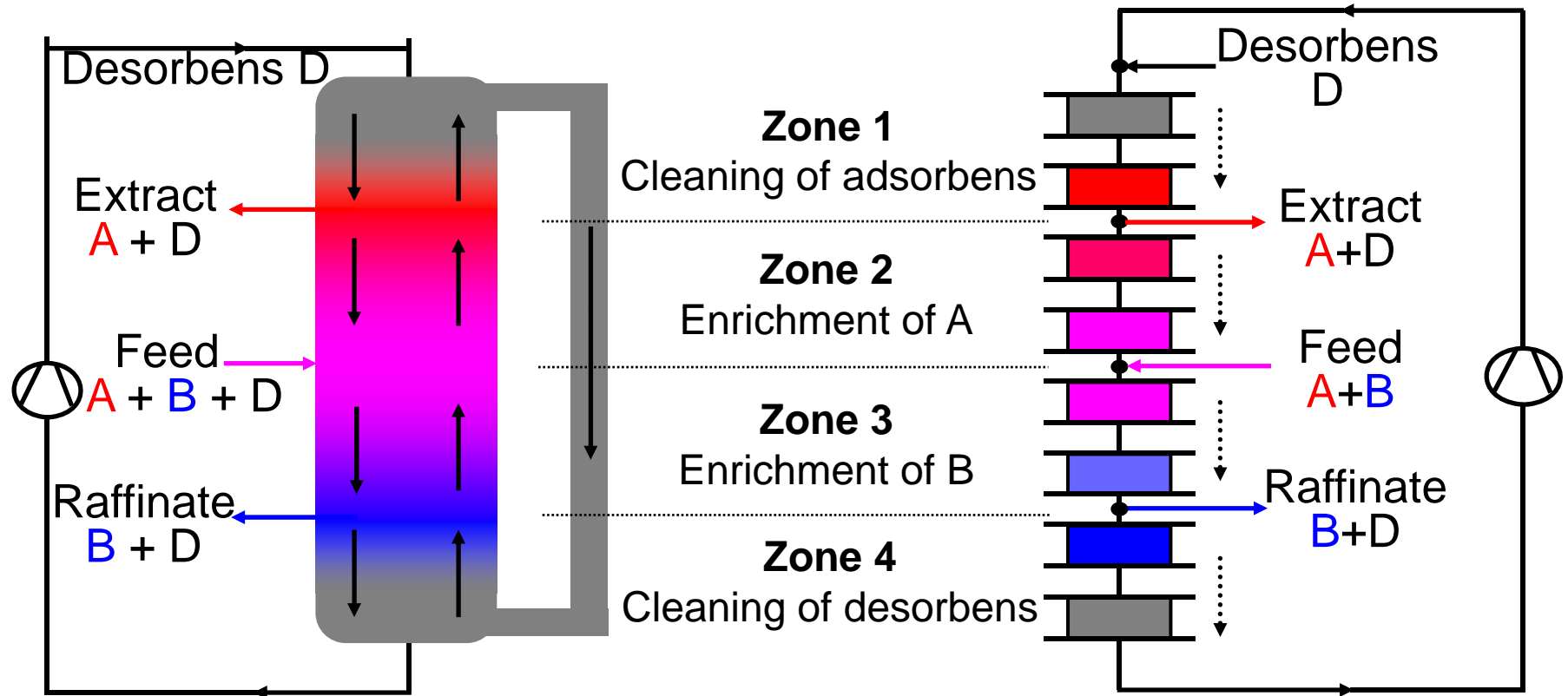


Preparative Batch-SFC



ID = 3 cm
 $L_{\max} = 45$ cm
 $P_{\max} = 40$ MPa
 $T_{\max} = 470$ K
max. 450 g adsorbent

Continuous chromatography



True Moving Bed (TMB)

Simulated Moving Bed (SMB)

Chromatography: Batch versus SMB

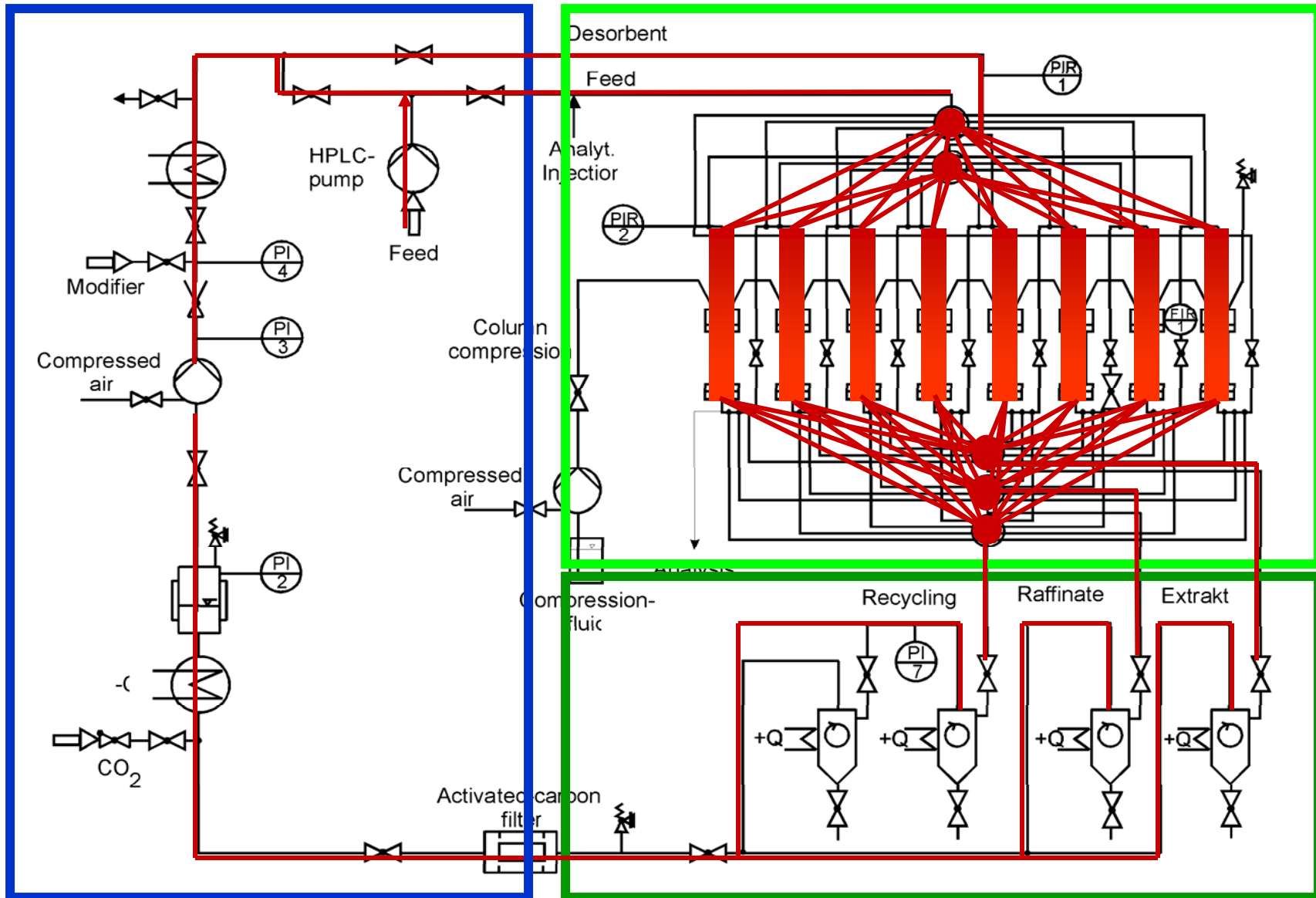
Batch-Chromatography

- ↓ discontinuity
- ↓ high solvent consumption
- ↓ dilution of products
- ↓ lost in yield due to waste fractions
- ↑ multi-component separations possible
- ↑ simple up-scaling

SMB-Technique

- ↑ continuous process
- ↑ lower solvent consumption
- ↑ higher concentrated products
- ↑ high yield (no waste fraction)
- ↓ especially for binary separations
- ↓ more complicated process development / technically demanding

SMB-SFC



SMB-SFC



ID = 3 cm

$L_{\max} = 19$ cm

$P_{\max} = 40$ MPa

$T_{\max} = 470$ K

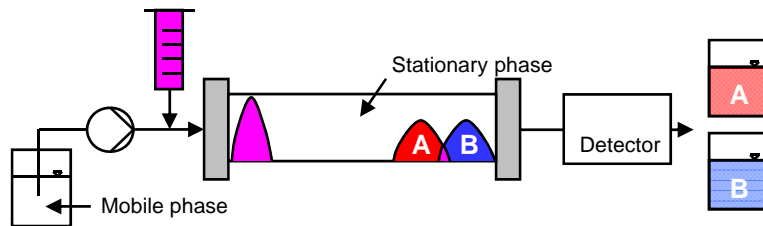
max. 8 x 100 g adsorbent

Preparative SFC: Applications at TUHH

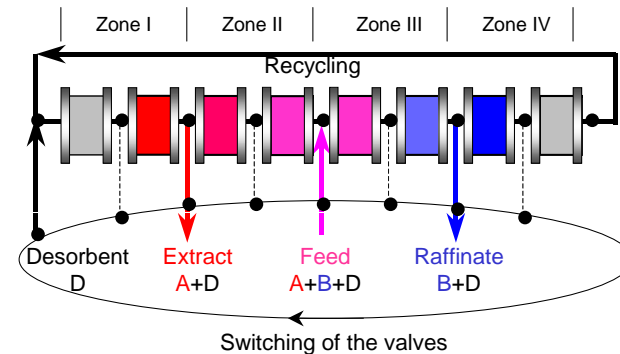
- ◆ **Prostaglandines** on LiChrosorb Si60
- ◆ **Fish oil ethyl esters (DPA/DHA)** on Alumina
- ◆ **cis-/trans-Phytol** on LiChrospher Si60
- ◆ **Vitamin D₃** on Zorbax Pro 10-60 CN
- ◆ **γ-/δ-Tocopherol** on LiChrospher Si60
- ◆ **R-/S-Ibuprofen** on Kromasil CHI-TBB
- ◆ **R-/S-Binaphthol** on Kromasil CHI-DMB
- ◆ **R-/S-Lacton** on Chiralpak AS
- ◆ **R-/S-1-Phenyl-1-propanol** on Chiralcel OD
- ◆ **Tocochromanols** on Kromasil KR60-10-SIL

Process development

Batch-SFC



SMB-SFC



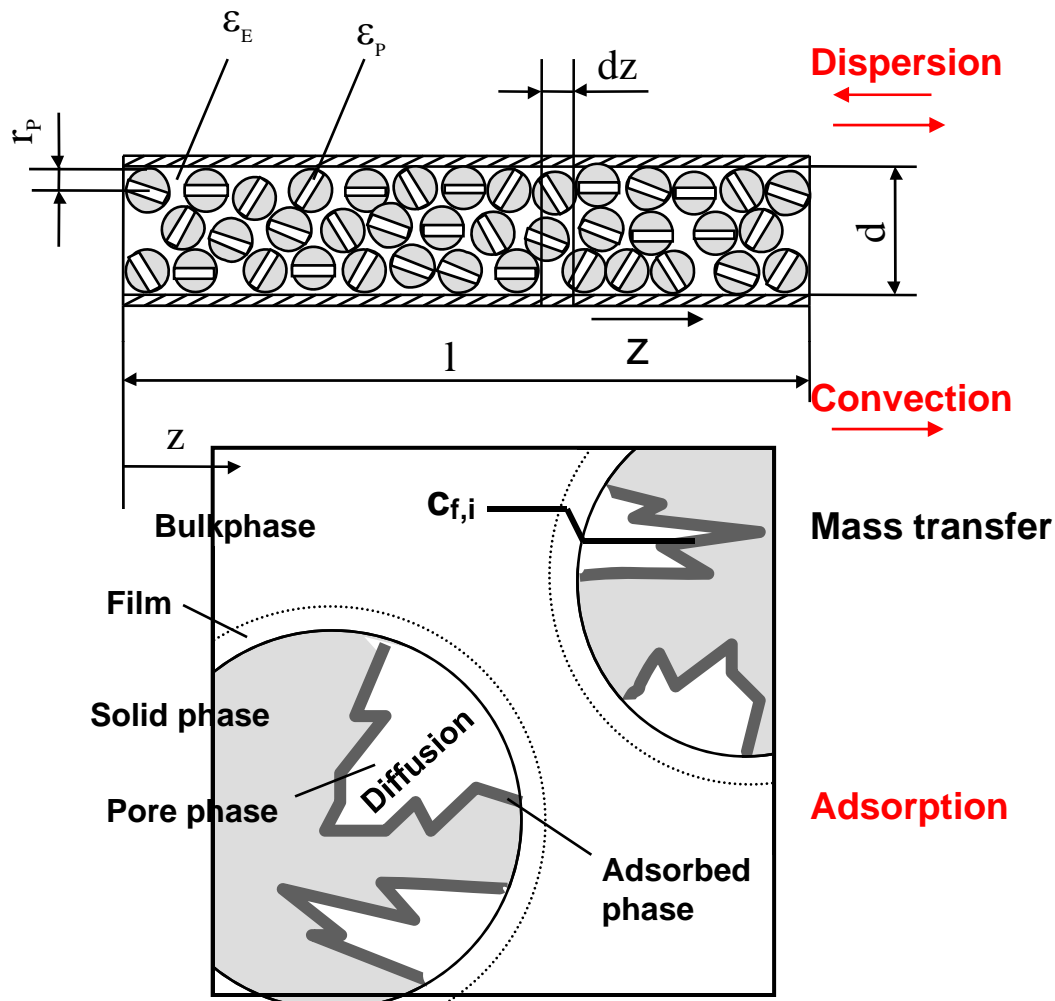
Parameters of influence:

- Separation system (adsorbent, particle size, mobile phase composition)
- Column geometry
- Operation parameter (pressure, temperature, flow rate, injection volume, fraction period)

Additional parameters:

- Column configuration
- Flow rates in the different zones
- Switching time of the valves

Model assumptions



Used balance equation:

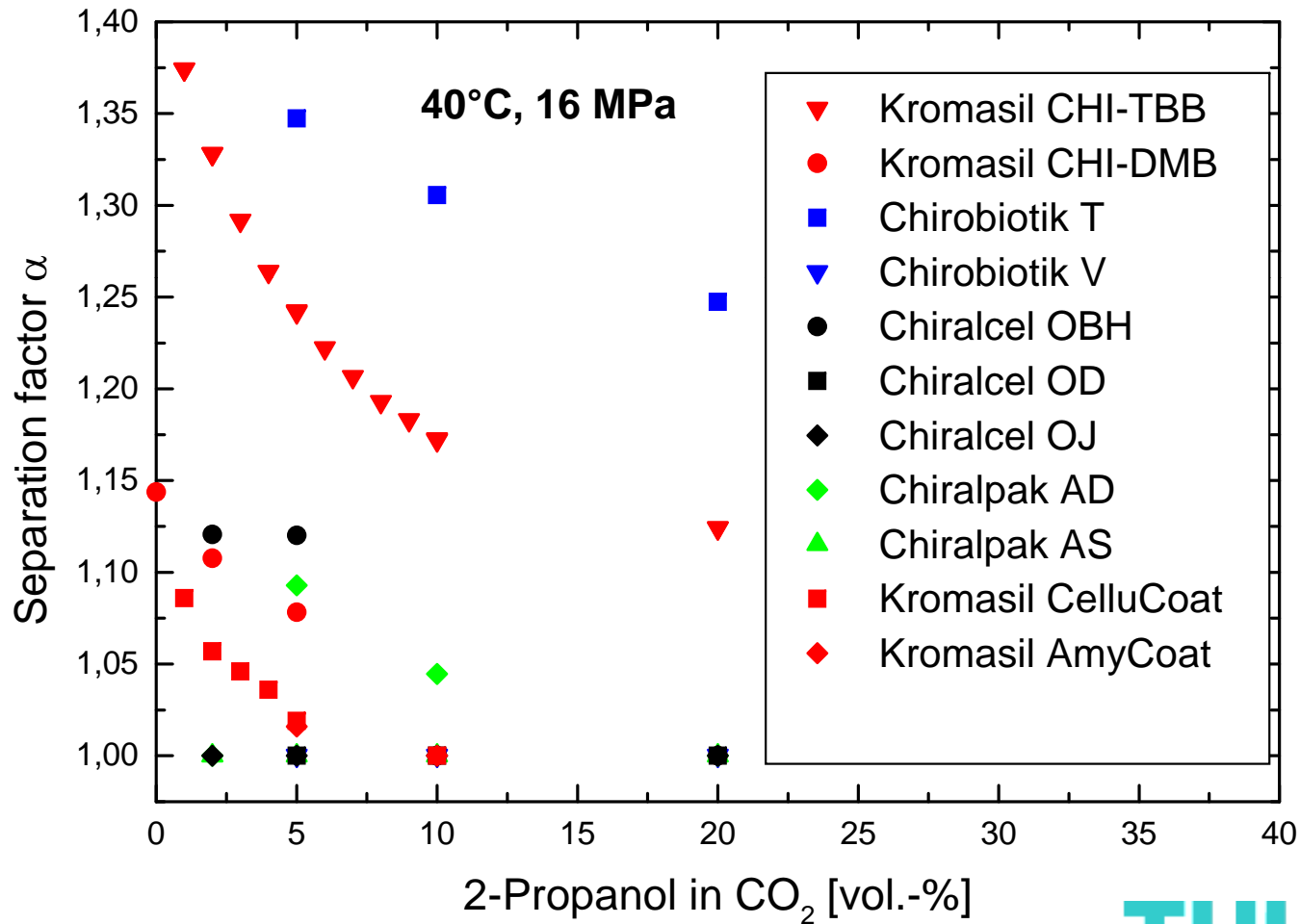
- Equilibrium dispersive model
- Mobile phase velocity is considered to be variable

Adsorption:

- Single component isotherms determined (cubic Hill isotherm)
- Describing binary adsorption by the ideal adsorbed solution theory (IAST)

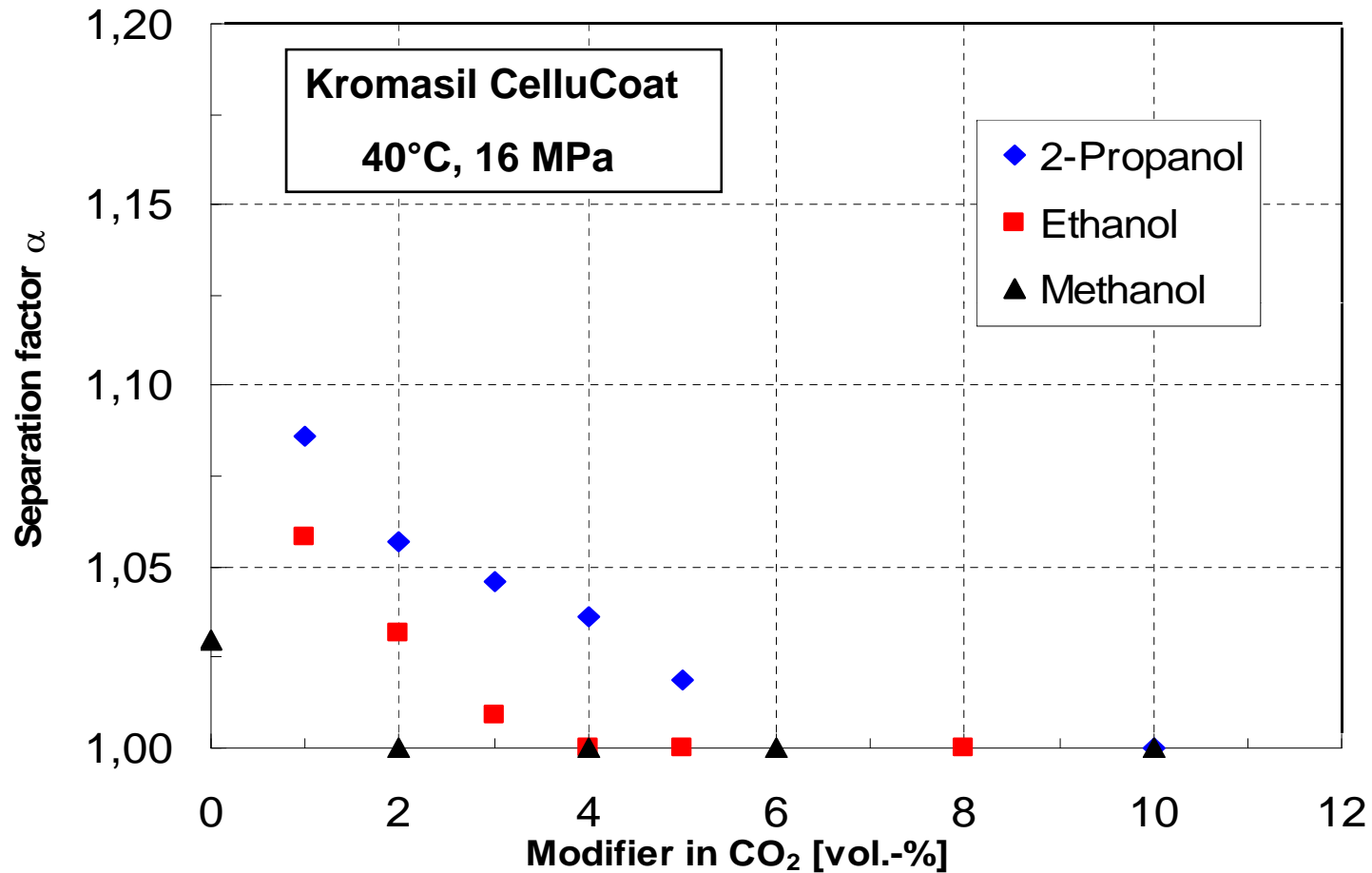
Selection of stationary phase

SFC separation of (R,S)-ibuprofen



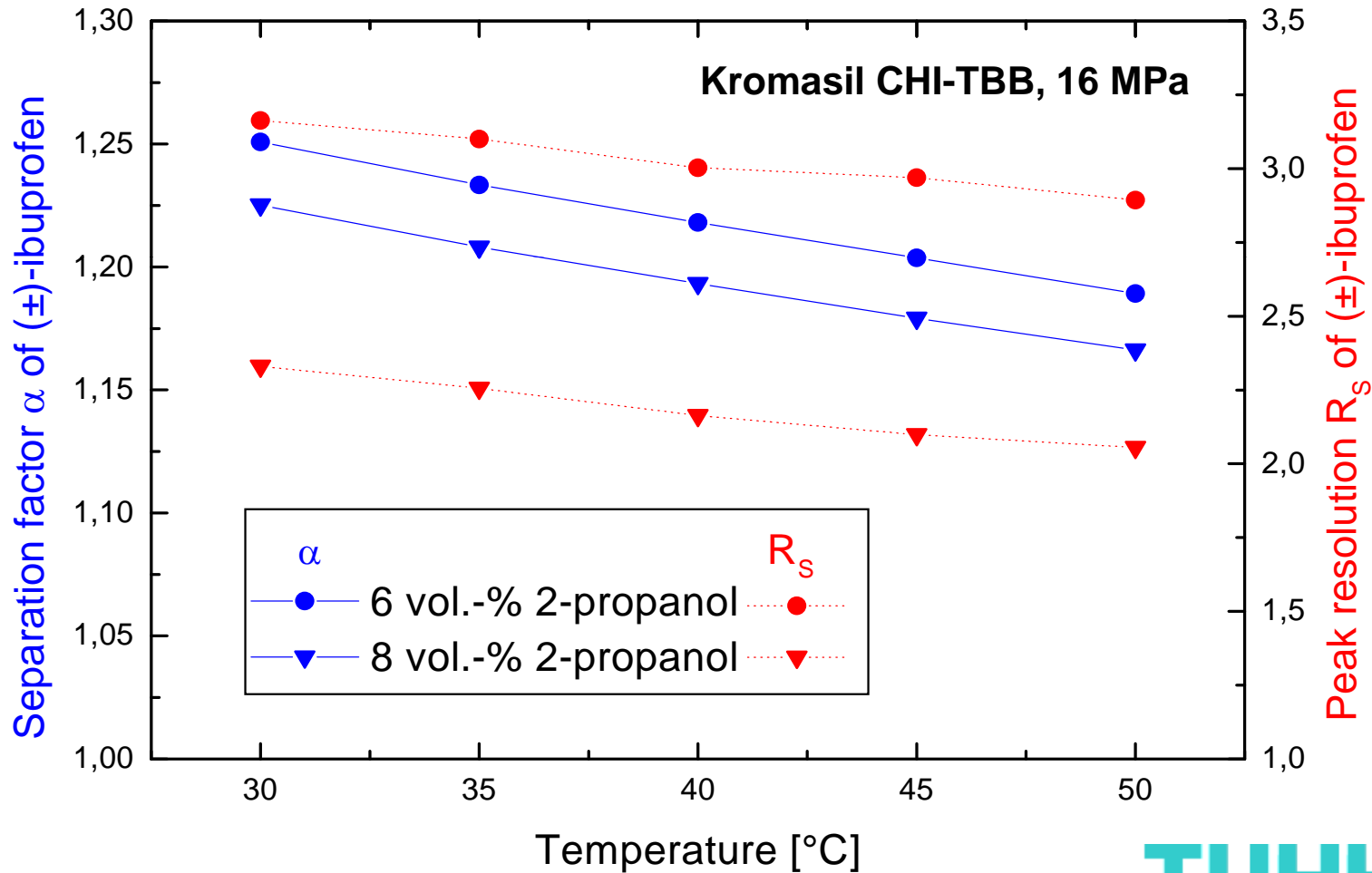
Selection of mobile phase

SFC separation of (R,S)-ibuprofen



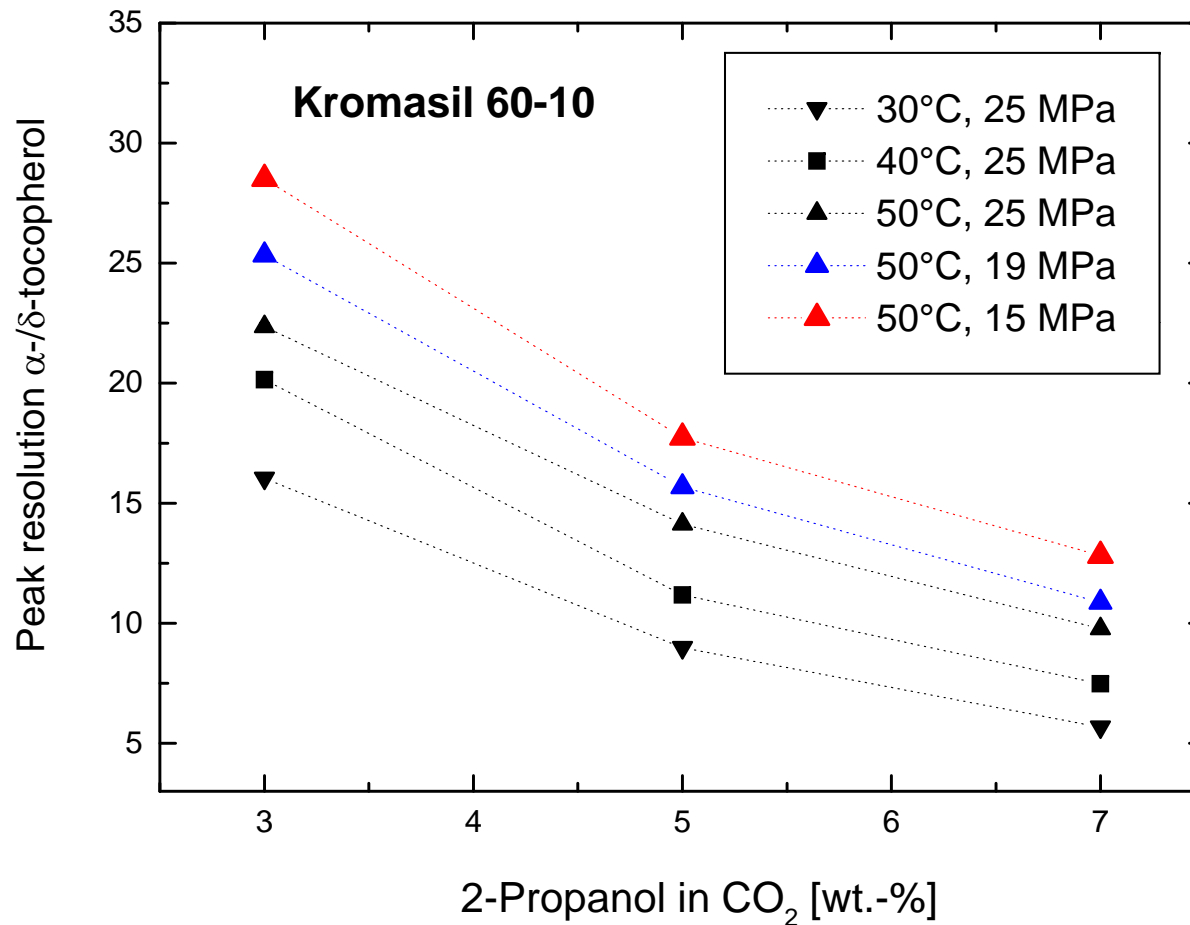
Influence of temperature

SFC separation of (R,S)-ibuprofen



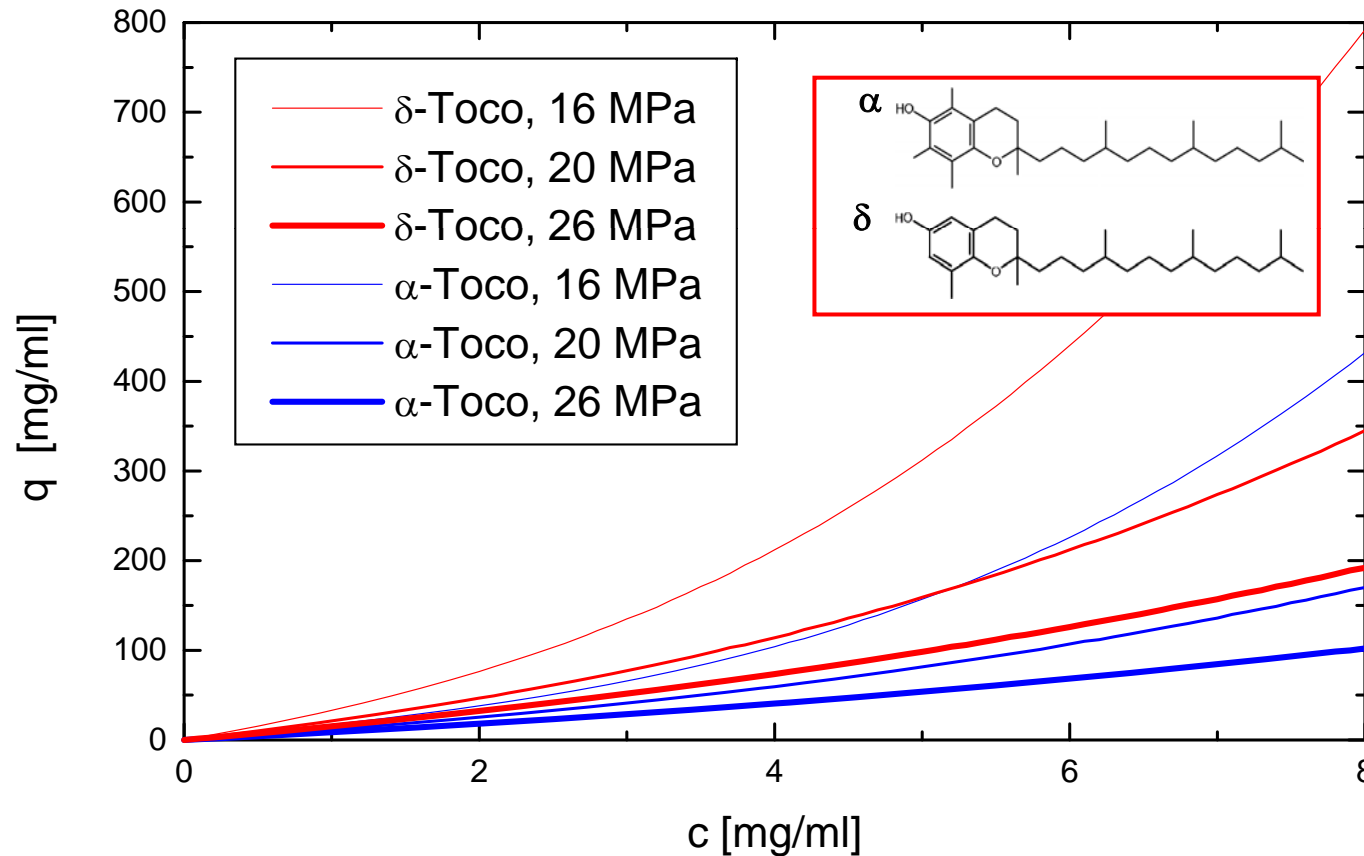
Influence of temperature and pressure

SFC separation of α -/ δ - tocopherol



Measurement of adsorption isotherms

Adsorption of α -/ δ -tocopherol on Kromasil 60-10, 313 K, 5% 2-Propanol in CO_2



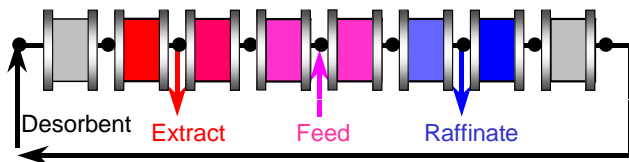
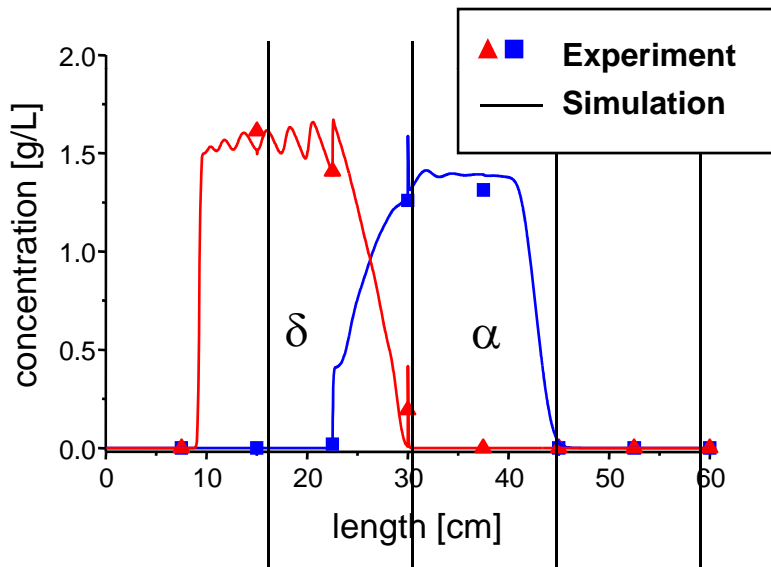
[Lübbert et al., 2006]

Comparison experiment – simulation

Separation of α -/ δ -tocopherol on Kromasil (5 wt.-% 2-propanol in CO₂)

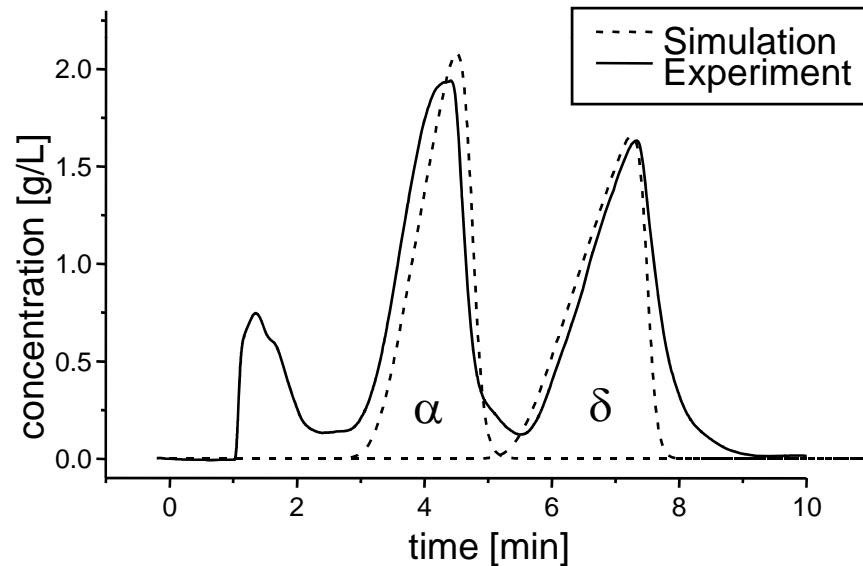
SMB-SFC

$C_{\text{Feed}} = 10 \text{ g/L}$; $p_m = 20 \text{ MPa}$; $T = 314 \text{ K}$



Batch-SFC

$C_{\text{Feed}} = 40 \text{ g/L}$; $V_{\text{inj}} = 10 \text{ ml}$; $p_m = 20 \text{ MPa}$;
 $T = 313 \text{ K}$; $L_{\text{col}} = 16.6 \text{ cm}$



Comparison experiment – simulation

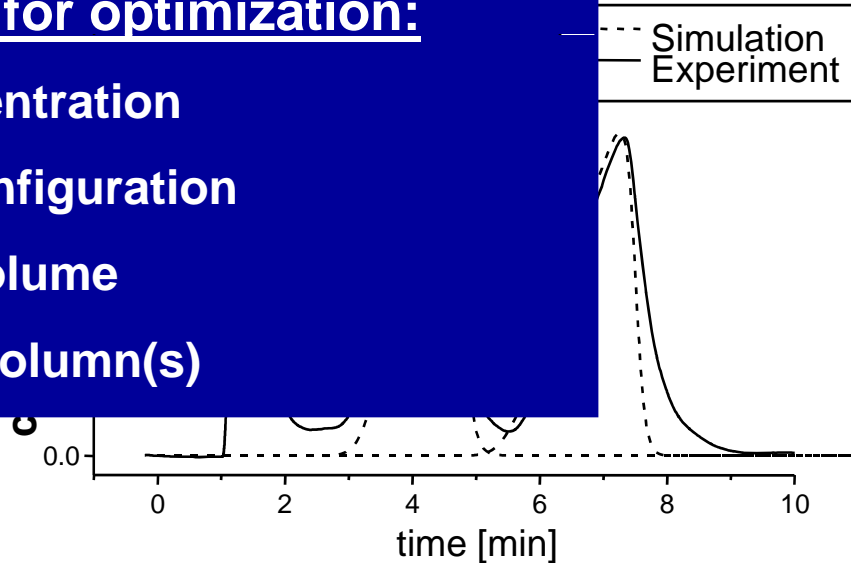
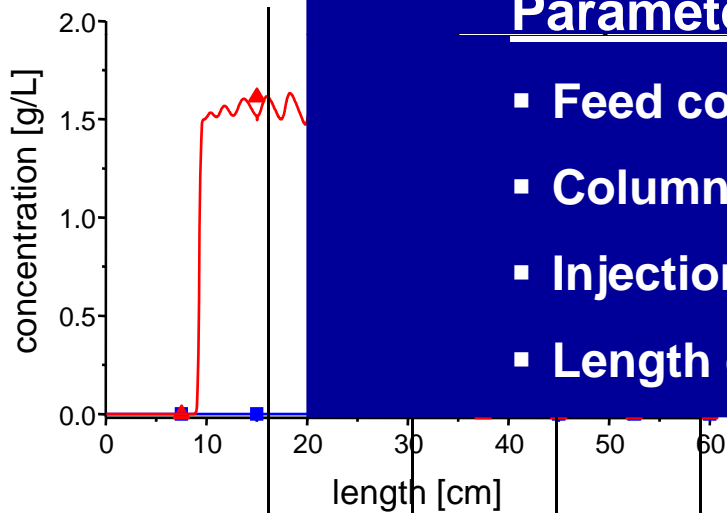
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SMB-SFC

Batch-SFC

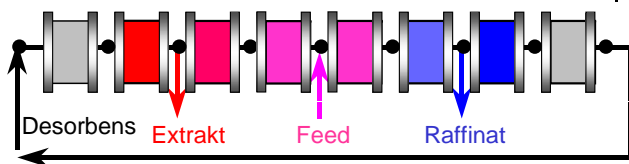
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 $T = 313 \text{ K}$; $L = 16.6 \text{ cm}$



Parameters for optimization:

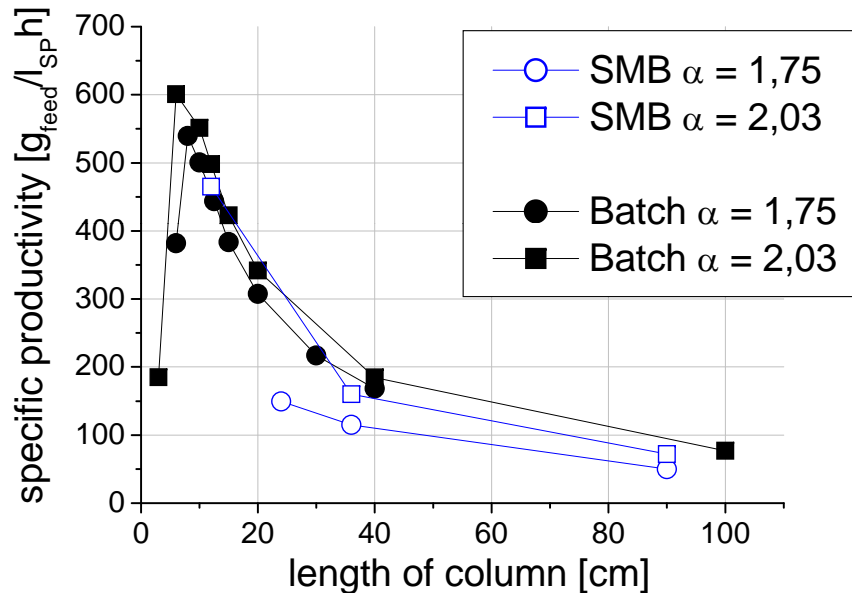
- Feed concentration
- Column configuration
- Injection volume
- Length of column(s)



Specific productivities

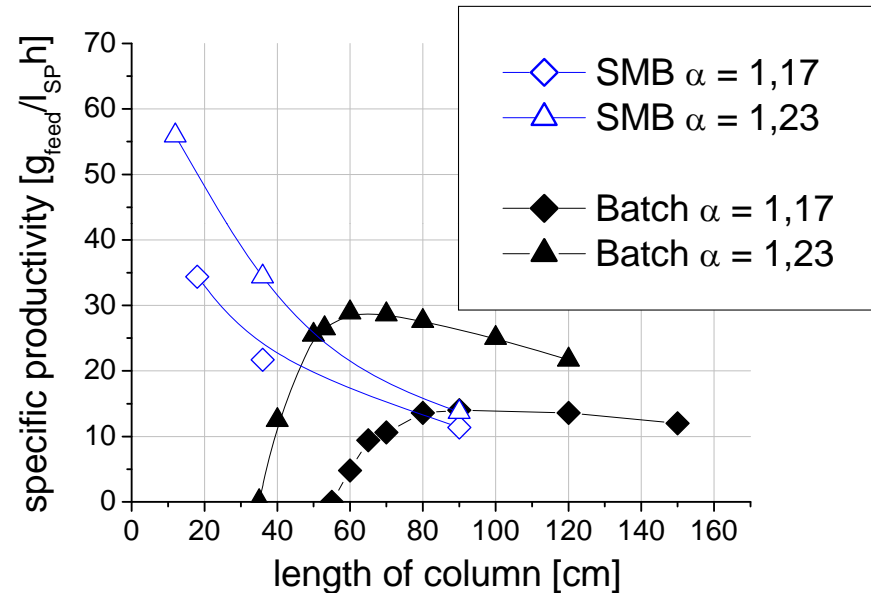
$N_{th} = 2500 \text{ m}^{-1}$, $u_{max} = 0.84 \text{ cm/s}$; SMB column configuration: 1/2/2/1

High separation factor



- **High separation factor:** Specific productivities nearly the same for both processes (Batch and SMB)

Low separation factor



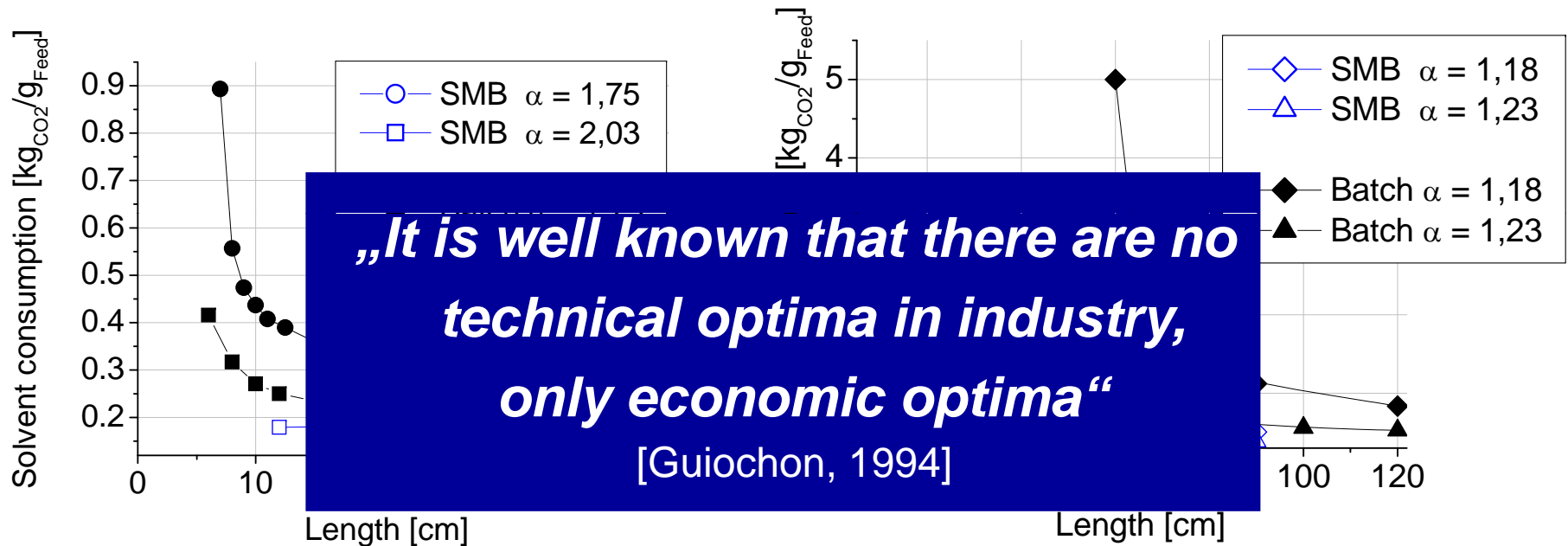
- **Low separation factors:** Specific productivities of the SMB-process significant higher

Solvent consumption

$N_{th} = 2500 \text{ m}^{-1}$, $u_{max} = 0.84 \text{ cm/s}$; SMB column configuration: 1/2/2/1

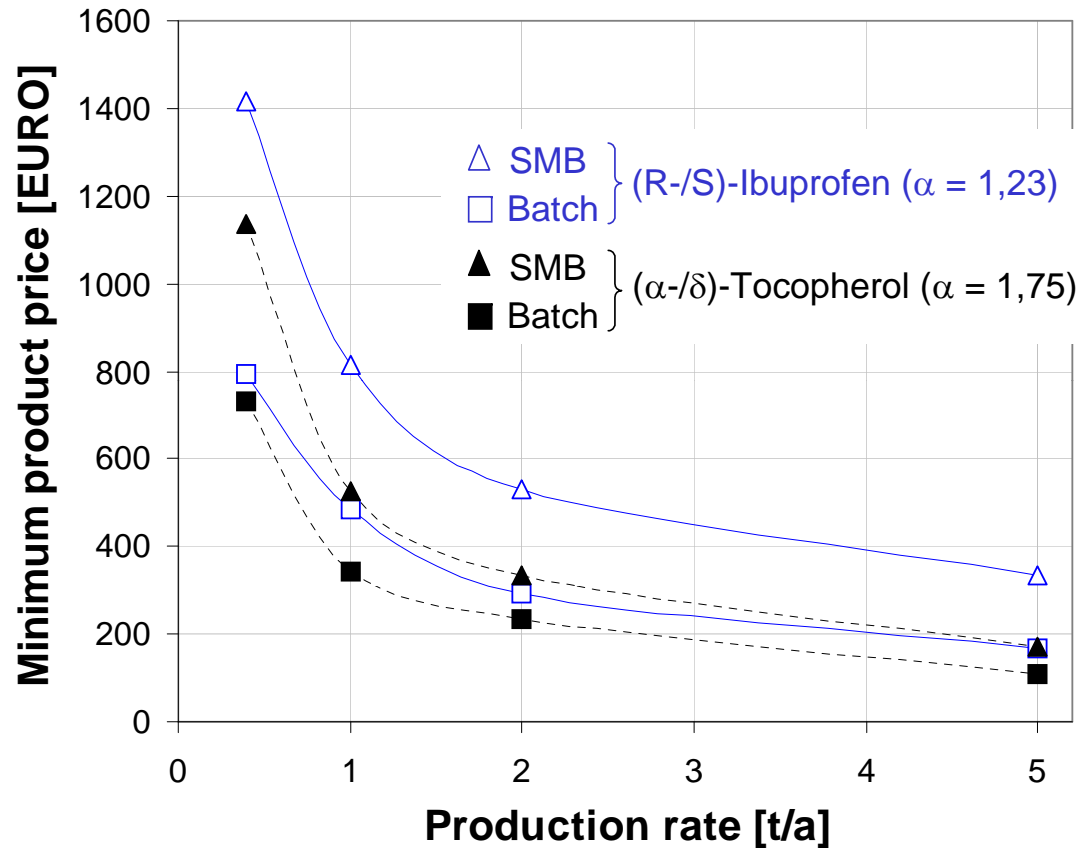
High separation factor

Low separation factor



- The solvent consumption depends on the separation factor
- Solvent consumption of SMB-process significant lower

Minimum product price



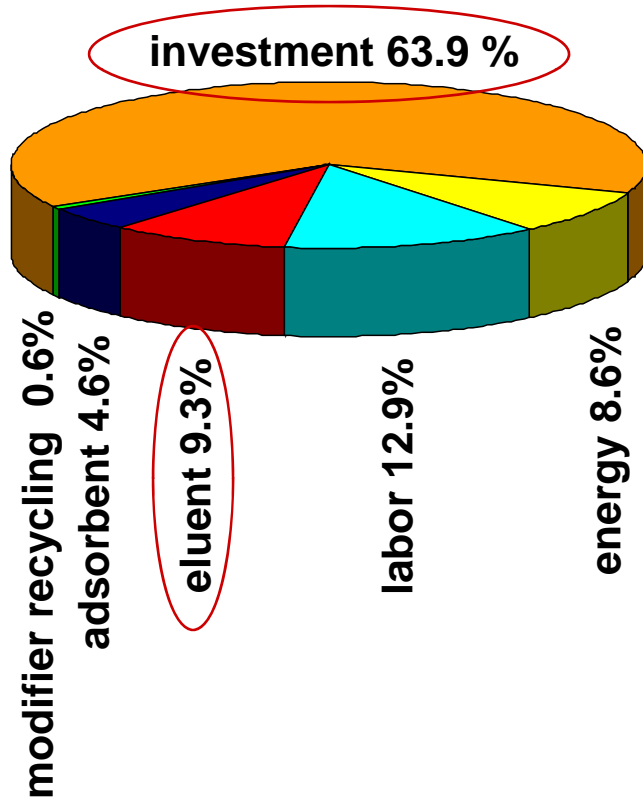
- In the range of investigated production rates, the Batch-process is the cheaper process

Distribution of costs

Separation of (R,S)-ibuprofen: 5 t/a (eluent: CO₂ + 2-propanol)

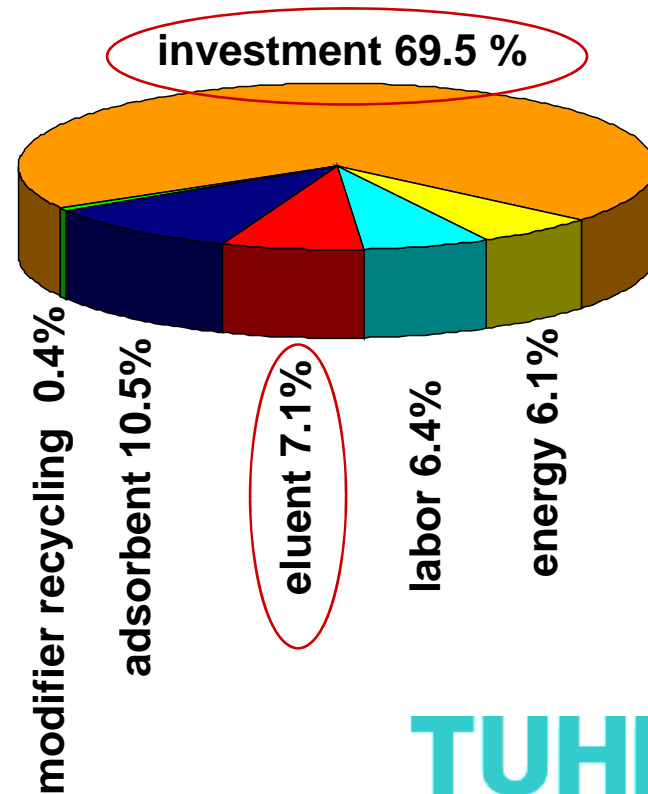
Batch-SFC

167 €/kg



SMB-SFC

337 €/kg



Summary

- Instruments for SFC at different scales are commercially available
- Batch-SFC (250 g st. phase) and SFC-SMB with up to 8 axial compressed columns (700 g st. phase) at TUHH
- Miscellaneous preparative SFC separations were carried out at TUHH successfully
- Adsorption equilibria as basis for optimisation of chromatographic processes
- Methods for process optimisation in SFC (Batch and SMB) are at hand
- In the range of the investigated production rates, the Batch-SFC is the more profitable process compared to SMB-SFC (cheap and easy!)

Summary

- Instruments for SFC at different scales are commercially available
- Batch-SFC (450 g st. phase) and SFC-SMB with up to 8 axial compressed columns (800 g st. phase) at TUHH

Supercritical fluid chromatography:

Ready for broad application as successful separation process in chemical and pharmaceutical industry!

- Methods for process optimisation in SFC (Batch and SMB) are at hand
- In the range of the investigated production rates, the Batch-SFC is the more profitable process compared to SMB-SFC (cheap and easy!)

THANKS

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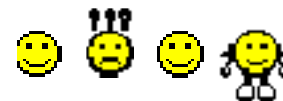
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